STUDY GUIDE FOR FUMIGATION

The educational material in this study guide is practical information to prepare you to meet the written test requirements. It doesn't include all the things you need to know about this pest-control subject or your pest-control profession. It will, however, help in preparation for your test.

Fumigants are quick acting and highly toxic to man and other organisms, requiring special safeguards in their use. Always read the label of a pesticide before any use. It is the legal document regarding its use, and the best source of information.

Contributors include the Utah Department of Agriculture and Utah State University Extension Service. This study guide is based on a similar one published by the Colorado Department of Agriculture. Materials were prepared by Colorado State University Extension Service. Other contributors include: Extension Service personnel of California, Illinois, and Georgia as well as the materials prepared in the previous draft by Metro-Pest Management Consultants, Inc. were utilized freely and with appreciation in preparing this study guide.

The information and recommendations in this study guide are based on data believed to be correct. However, no endorsement, guarantee or warranty of any kind, expressed or implied, is made with respect to the information contained herein.

Other topics that may be covered in your tests include First Aid, Personal Protective Equipment (PPE), Protecting the Environment, Pesticide Movement, Groundwater, Endangered Species, Application Methods and Equipment, Equipment Calibration, Insecticide Use, Application, Area Measurements, and Weights and Measures. Information on these topics can be found in the following books:

- 1. Applying Pesticides Correctly: A guide for Private and Commercial Applicators. U.S. EPA, USDA and Extension Service, revised 1991.
- 2. Applying Pesticides Correctly: A Supplemental Guide for Private Applicators. U.S. EPA, USDA and Extension Service, December 1993, Publication E-2474.

These books can be obtained from the Utah Department of Agriculture or Utah State University Extension Service. Please contact your local Utah Department of Agriculture field representative or Utah State University extension agent.

TABLE OF CONTENTS

FUMIGATION

INTRODUCTION	1
HOW FUMIGANTS WORK	1
HOW TO CHOOSE A FUMIGANT	4
TYPES AND NATURE OF FUMIGANTS	5
DETERMINING A NEED FOR FUMIGATION	5
TARGET-PEST CONSIDERATIONS	5
SITE-SUITABILITY CONSIDERATIONS	6
FACTORS AFFECTING FUMIGANT PERFORMANCE	7
SAFETY PRECAUTIONS AND PROTECTIVE DEVICES	9
CALCULATION USE RATES	12
SEALING	13
APPLYING THE FUMIGANT	14
POSTING AND SECURING A FUMIGATION SITE	15
AERATING	15
SPECIAL CONSIDERATIONS FOR FUMIGANT AND CONTAINER DISPOSAL	16
THREATENED AND ENDANGERED SPECIES	19
WORKER PROTECTION STANDARDS	19
GROUNDWATER CONTAMINATION BY PESTICIDES	20

INTRODUCTION

A fumigant is a chemical vapor or gas that, when released, penetrates objects or enclosed areas in concentrations that are lethal to pest organisms. This definition excludes aerosols, which are particles suspended in the air, often referred to as smokes, fogs or mists. It's important to make this distinction, since it emphasizes one of the most important and useful properties of fumigants: as gases, they diffuse as separate molecules. This allows them to penetrate into the material being fumigated and to diffuse away afterward.

Some insecticides, when sprayed on leaves or other surfaces as contact or stomach poisons, sometimes give off a gas. This gas may account for part of the toxic action of these applications. This is called the fumigation effect. This study guide won't deal with this effect; the guide is limited to fumigants that are dispensed so that the poison is present as a gas soon after application and reaches the pest as a gas.

Fumigation techniques have great adaptability in pest control. They can be used to control wood-destroying insects in structures and furniture where liquid or dust formulations are ineffective or where these materials may cause damage. Under some conditions, fumigants can be applied to control burrowing rodents that can't be reached with other types of rodenticides. Most commonly, fumigants are used in Utah to control insects or mites in fresh and stored food products such as grains, fruits, vegetables, nuts and dried fruit. Fumigation may take place at a home or storage facility, or it may occur in a carrier, such as a truck or railway car.

Controlled-atmosphere storage of certain food products is a unique form of fumigation. In a controlled atmosphere, most of the air in an enclosed storage area is replaced with a gas such as carbon dioxide.

Before performing a fumigation, the applicator needs to understand clearly the hazards and problems associated with the use of fumigants. Most fumigants are highly toxic to all forms of life, including humans, animals, plants, and even microbes. Fumigation is a highly specialized operation that requires equipment, techniques, and skills not generally used for applying

other types of pesticides. Applying a fumigant may be time-consuming and expensive, usually requiring more labor than other pest-control methods.

Structural fumigation is disruptive, since it requires that tenants and other occupants leave the building. Because of the special hazards and conditions of fumigation, strict legal restrictions exist concerning its use.

HOW FUMIGANTS WORK

Fumigants kill by interfering with the respiratory function of the target pest. Molecules of some fumigants (for instance, carbon dioxide or inert gases) replace oxygen molecules in the air, so the pest-control action involves smothering (asphyxiation) due to lack of oxygen. Other fumigants enter tissues and disrupt enzymes used in the respiration of animal or plant cells.

The killing action of a fumigant is influenced by its concentration in the atmosphere, the length of time it stays in the atmosphere, and the temperature and humidity of the area at the time of fumigation. Fumigants are designed to enter cracks, crevices and other areas where target pests may occur. They must be applied in enclosed areas. Fumigation has no residual effect, and re-infestation may occur after the fumigant has diffused from the area.

Advantages of Fumigation

Fumigation has several advantages over other methods of pest control:

- ! Fumigants are usually quick-acting and can result in total eradication of the pest.
- ! Because fumigants are gases, they diffuse through all parts of the structure or commodity being treated and can reach pests that couldn't be reached with conventional pest-control materials or techniques.
- ! For certain commodities, fumigation is the only practical way to control pests.

Disadvantages of Fumigation

There are several reasons why fumigation sometimes may not be the best means of pest control. These are:

- ! The control achieved through fumigation is temporary -- there's no residual action from fumigants. Where untreated populations of the pest remain, re-infestation of the treated site can take place quickly.
- ! Fumigants are toxic and often highly hazardous to the applicator, requiring special precautions during application.
- ! Fumigants must be retained in the gas form for a period of time to be effective, often calling for extra supervision.
- ! Fumigation must never be done by just one person, which requires added labor.
- ! Some commodities or pieces of equipment may be damaged by certain fumigants and must be removed or otherwise protected.
- ! Fumigant activity may be greatly affected by temperature and humidity.

HOW TO CHOOSE A FUMIGANT

If the need for fumigation has been proven, the right fumigant must be chosen. To decide on an effective fumigant, it's important for the applicator to know the habits of the pest, the characteristics of the fumigation site, and environmental conditions that may influence the fumigation process. He or she also needs to understand the chemical and physical characteristics of the fumigant.

When choosing a fumigant, consider such factors as these:

- ! Toxicity to the target pest
- ! Volatility and ability to penetrate
- ! Corrosive effect, flammability, and potential for explosion
- ! Warning properties and detection methods
- ! Effect on seed germination and finished-product quality
- ! Residue tolerances
- ! Availability
- ! Ease of application
- ! Cost.

CHEMICAL AND PHYSICAL CHARACTERISTICS

Important physical and chemical characteristics of a fumigant include volatility, molecular weight, boiling point, vapor pressure, specific gravity, diffusion potential, water solubility, latent heat of vaporization, flammability, and chemical reactivity. **Read the product information** supplied by the manufacturer to be sure that the material you select is appropriate to the commodity, treatment site, and pest control needs. The label must list the commodity and target sites.

Volatility

Volatility is the tendency of a chemical to evaporate and become a gas or vapor. Volatility increases as temperature rises. Some "gaseous-type" fumigants, such as methyl bromide, are normally a gas at room temperature. Other fumigants exist as a liquid or solid (paradichlorobenzene, naphthalene) at room temperature. Also, many of the "solid-type" fumigants, such as aluminum and magnesium phosphide, are not fumigants themselves but react with moisture to form a fumigant gas (phosphine or hydrogen phosphide).

Molecular weight

Molecular weight is a measure of the weight of the atoms that form the fumigant molecule. More complex molecules have greater molecular weight because they have more atoms. Larger molecules are often less suitable as fumigants, since they are less volatile.

Boiling point

The boiling point of the chemical is the temperature at which the liquid stage boils under specific atmospheric conditions to become a gas. Some materials used as fumigants, such as methyl bromide, have low boiling points so they are gases at normal temperatures and atmospheric pressure. These types of fumigants are usually stored as liquids under high pressure.

The boiling point of a fumigant may influence the type of application equipment required. For example, fumigants with low boiling points usually require heaters to warm the gas as it's being released. This is because these materials may freeze on release into the atmosphere,

since much heat is lost as fumigants turn from a liquid to a gas.

Vapor pressure

The vapor pressure of the fumigant affects the atmospheric concentration of the gas in the air. When a volatile liquid or solid is confined in an area, equilibrium gradually takes place between molecules in the gas and liquid phases. Once the gas molecules reach a saturation point, further volatilization won't increase the number of molecules in the vapor phase. Although volatilization may appear to stop, what actually happens is that every molecule evaporating from the liquid is replaced by a gas molecule condensing back to the liquid form. Since vapor pressure determines the concentration that can be maintained during fumigation, materials of high vapor pressure will be more concentrated and therefore have better fumigant qualities.

Specific gravity

The specific gravity of a chemical compound is a measure of its weight in a given volume. With fumigants, it's important to know if the gas is lighter or heavier than air. Most commonly used fumigants are heavier than air. A heavy gas in a confined area will tend to concentrate in low areas and mix slowly with the air.

These fumigants usually require mechanical mixing with a fan to distribute the molecules evenly through the fumigated area. However, once the fumigant is thoroughly mixed with the air, settling takes place very slowly. As a result, the problem of stratification -- or layering -- of heavier-than-air fumigants doesn't have much practical meaning for the exposure periods usually required in fumigation work.

All gases become lighter as they become warmer. This is because warm molecules take up more space, so fewer molecules can be contained in a given space at the same pressure.

Diffusion potential

Diffusion potential is a measure of how fast gas molecules disperse through the atmosphere. After a while, the molecules become evenly distributed. The speed with which molecules disperse is affected by the molecular weight of the gas. Gases that are heavier

diffuse more slowly, and it may be important to disperse these types of gases with fans or blowers.

Water solubility

The water solubility of a fumigant becomes an important consideration if items in a fumigated area contain even small amounts of water. The water will tie up water-soluble fumigant molecules, reducing the fumigant concentration in the atmosphere. Toxic molecules also may be incorporated into the water of fumigated materials and may remain as undesirable residues. Suitable fumigants for most applications are those that are insoluble or only slightly soluble in water.

Latent heat of vaporization

Latent heat of vaporization (the extra heat required to change the liquid to a gas) must be considered when using fumigants that have boiling points below room temperature. Unless sustained by warming from an outside source, the temperature of an evaporating liquid constantly drops. This is shown by the cooling effect of evaporating water on the skin.

The factor of latent heat has important practical significance. High-pressure fumigants, such as methyl bromide, volatilize and lose heat rapidly on release. Unless the lost heat is restored, the temperature of the fumigant may fall below its boiling point, causing the gas to no longer evolve. Also, as the liquid changing to gas is led through metal pipes and tubes or rubber tubing, the drop in temperature may freeze the fumigant in the lines, preventing further passage.

In many applications, it's wise to apply heat to the fumigant as it passes from the container into the fumigation space. Fumigants that are liquids at normal temperatures and are volatilized from evaporating pans or vaporizing nozzles also lose heat. These applications may require a source of heat, such as a hot plate, so that full concentrations will take place rapidly.

Flammability

Flammability of a fumigant is another physical characteristic that is very important in its safe use. Fumigants that are flammable gases are usually combined with a non-flammable gas (such as carbon dioxide) to reduce the danger of fire or explosion.

Chemical reactivity

Chemical reactivity of some fumigants with other chemicals in the environment may limit some fumigant uses. For example, methyl bromide combines with sulfur-containing compounds (such as rubber, leather and other animal products) and gives off a strong, foul odor that is hard to eliminate. Phosphine gas reacts with copper (used in electrical wiring, motors and plumbing) to cause serious corrosion. High temperatures around an open flame may cause some fumigants to form corrosive acids. Certain fumigants may make photographic film and paper unusable because of chemical reaction.

TYPES AND NATURE OF FUMIGANTS

Many of the active ingredients in fumigants used earlier have either been canceled entirely or had their uses restricted. All space-fumigation products and several soil- fumigant products (especially those containing chloropicrin and/or methyl bromide) are now restricted-use pesticides.

Active ingredients that are still legal to use include:

- 1. Methyl bromide
- 2. Chloropicrin
- 3. Aluminum phosphide
- 4. Magnesium phosphide
- 5. Sulfuryl fluoride
- 6. Carbon dioxide

Methyl Bromide

Methyl bromide readily penetrates many materials and is in wide use for space fumigation. Methyl bromide is also used in agriculture as a soil fumigant to control fungi, weeds, nematodes and insects. Methyl bromide is sold as a liquid under pressure. Upon release, it vaporizes to form a gas that is about 3.3 times heavier than air.

Methyl bromide is a colorless, odorless and tasteless gas, but it's highly toxic as a respiratory poison and can cause serious eye and skin damage. It's usually formulated with a small amount of chloropicrin as a warning agent. Early symptoms of overexposure are dizziness, headaches, nausea and vomiting, weakness, and collapse. Fluids in the lungs and heart irregularities may develop two to 48 hours after exposure. These effects can result in death.

Methyl bromide is retained, at least for a short time, in body tissues. Repeated small over exposures can cause symptoms such as blurred vision, staggering walk, and mental imbalances, with probable recovery after a period of no exposure. If trapped inside tight clothing next to the skin, methyl bromide can cause severe skin burns.

Methyl bromide reacts chemically with sulfur products and should not be used to fumigate materials such as fur, leather, rubber, wool, and feathers.

Chloropicrin

Chloropicrin fumigants include products marketed under the names Chlor-O-Pic, Lavacide 100 and Quasar. These products contain nearly 100-percent chloropicrin and are marketed as liquids. Chloropicrin volatilizes to form a dense gas that is about 5.7 times heavier than air.

Chloropicrin is highly toxic to insects, vertebrates, and many soil microbes, such as fungi. It's highly irritating to eyes and is a powerful "tear gas." Concentrations as low as 1.0 parts per million (ppm) cause intense eye irritation, and prolonged exposures cause severe lung injury. Chloropicrin can cause severe injury upon skin contact. Uses of chloropicrin on foodstuffs have been restricted in recent years. Right now, use is prohibited on most food, and direct-grain treatment uses are under review.

Aluminum Phosphide

Aluminum-phosphide fumigants include products marketed under the trade names Detia, Fumitoxin, Gastoxin, Phostek and Phostoxin. These products contain aluminum phosphide in combination with inert ingredients such as ammonium carbamate and urea. The formulated material is a solid molded into pellets or tablets. The active ingredient, aluminum phosphide, reacts with atmospheric water to produce hydrogen-phosphide gas. This gas is also known as phosphine. Phosphine is a colorless gas with an odor that smells different to different people. The odor is often described as similar to garlic, commercial carbide, or decaying fish.

Phosphine is only slightly heavier than air, about 1.2 times as heavy. Fumigators can't rely on the gas moving through a solid storage such as a grain bin, so they need to set up one or more fans to mix the fumigant with the air.

Aluminum phosphide is used commonly to fumigate grain-storage facilities. Phosphine gas is highly toxic to all forms of animal life. Early symptoms of poisoning can be severe, but these symptoms are reversible if exposure stops. Initial symptoms of overexposure include "tightness" in the chest, faintness, dizziness, nausea, vomiting and diarrhea. Severe poisoning leads to coma and death. Phosphine (hydrogen phosphide) gas isn't absorbed through the skin and it's not stored in body tissues. Aluminum phosphide may explode on contact with water.

Hydrogen-phosphide (phosphine) gas is reactive and very corrosive to metals, especially copper, silver, gold, and platinum.

Magnesium Phosphide

Magnesium phosphide is similar to aluminum phosphide, releasing hydrogen-phosphide gas in reaction with water. Release of the gas is faster than occurs with aluminum phosphide.

Common magnesium-phosphide products contain the solid magnesium-phosphide material attached to a strip or blanket that can be put in place very quickly. Because this application method may not provide good distribution of the gas in a grain mass, it isn't usually used in grain-storage fumigation. Magnesium-phosphide fumigants can be used effectively for warehouse and processing-plant fumigations.

Sulphuryl Fluoride

Sulphuryl fluoride, sold under the trade name Vikane, is a colorless, odorless gas. It's sold in canisters as a liquid under pressure that volatilizes readily. It's non-corrosive and unreactive to most materials. Sulphuryl fluoride also can provide good penetration of wood products and fabrics, but it needs fans or blowers to mix well with air. Sulphuryl fluoride is non-flammable, but in the presence of an open flame, it forms a very corrosive gas. It's highly toxic to humans.

Carbon Dioxide

Carbon dioxide (CO₂) is a colorless, odorless and tasteless gas that is about 1.5 times heavier than air. It's non-combustible and is used as a fire-extinguishing material. It's usually found in the air at concentrations of about 0.03 percent. However, carbon dioxide is poisonous at higher concentrations and is used for fumigating food products at about 60-percent concentration.

Using carbon dioxide is desirable because no toxic residues stay in treated materials. Also, CO₂ doesn't change the germination potential of treated grain and leaves no objectionable odor or flavors. However, fumigation with carbon dioxide requires fairly long exposure periods to be effective. Effectiveness is greatly reduced by low temperatures, so if temperatures are below 60 degrees F., fumigation periods may be too long to be practical (three to four weeks or more).

DETERMINING A NEED FOR FUMIGATION

Several criteria should be considered in determining the need and suitability of fumigation for pest control. These include:

- 1. Characteristics and habits of the pest
- 2. Life stages of the pest
- 3. Characteristics of the treatment area
- 4. Hazards located in the treatment area
- 5. Available pest-management alternative
- 6. Established pesticide-residue tolerances

TARGET-PEST CONSIDERATIONS

Fumigants used in pest control tend to affect all forms of life. Almost any pest in an enclosed area can be destroyed when exposed to an adequate concentration of a fumigant. Fumigations are most often used to treat pests that infest harvested commodities such as bulk grain, greenhouse insects, etc. Inaccessible pests, such as wood-boring beetles and drywood termites, are also targets for fumigation. Fumigation may sometimes be the best choice for controlling heavy infestations of

insects such as cockroaches, especially when it's hard to gain access to all of the pest's hiding places. Fumigation is also useful to avoid toxic residues associated with application of other pesticide formulations to food, clothing, and similar materials.

Habits of the Pest

Pests that are reclusive or hard to locate can often be treated successfully with fumigation. However, it's important to understand the habits of the pest before choosing fumigation. For example, colonies of drywood termites, very uncommon in Utah, nest in structural wood above ground and are good targets for fumigation. The far-more-common subterranean termites nest underground and are not killed by fumigation. Situations where large reservoirs of the pest will remain outside the treated area can allow quick re-infestation, wiping out the benefits of fumigation.

Life Stages of the Pest

An applicator should also consider how various life stages of the pest respond to fumigation. For instance, many insects are relatively non-susceptible to fumigants or other insecticides during their egg and pupal stages. Insects may also be dormant during certain periods and not be susceptible. Be sure to check the fumigant label to see what stages of the target pests the manufacturer claims the product will control.

SITE-SUITABILITY CONSIDERATIONS

Fumigation may be used in several types of situations, including structures, bulk-storage facilities, specially designed chambers, rail cars and trucks.

However, fumigants should only be used in enclosed areas, because the molecules of the fumigant penetrate throughout the area and escape through openings. Fumigants can't be used in localized areas of a building unless it's possible to completely seal and control access to the treated area throughout the fumigation and aeration period. Fumigants should never be used in any areas that can't be fully secured to prevent entry or contact by people or animals.

The fumigation site also must have the proper environmental conditions to allow successful use of the fumigants. This includes correct temperature, humidity, and air-circulation conditions required for effective pest control.

Sites should also be thoroughly surveyed to identify and protect items that may react with or be damaged by the fumigant. This may include such items as furnishings, floor coverings, foodstuffs, wall hangings, finishes, plumbing and electrical devices, and moisture sources.

Structural Fumigation

Fumigation may be used to control certain pests within existing buildings such as grain-storage bins and homes. Since typical construction isn't sufficiently airtight, these require sealing. In relatively airtight structures, taping may be sufficient. However, many buildings require tarping the entire structure.

Fumigants used in grain storage are very useful for control of stored-product insects such as weevils and various "bran bugs." Household fumigations can help control pests such as drywood termites, powder-post beetles, and other wood-boring beetles that are hard to control with other methods.

Chamber Fumigation

Since environmental conditions can be carefully controlled and monitored, chamber fumigation is a superior method for fumigating many materials. Using a chamber will allow only small amounts of a commodity to be fumigated at a time because of the limited size of the chamber. However, the limited space can be an advantage, because the fumigant is confined, saving the time it takes to fumigate and the amount of fumigant used. The ability to carefully control environmental conditions in a chamber also allows fumigation to be used to control pests on fragile commodities such as fresh fruits or vegetables without damage.

Chamber fumigation can also be used to disinfect fresh produce, packaged foods, bagged or baled agricultural products, museum specimens, furniture, high-value garments, and similar items.

Chambers used for fumigation may be either the atmospheric or vacuum type. Vacuum chambers

provide the quickest and most thorough fumigation and are best for finely divided items, such as flour. Applying a vacuum increases the penetration of a fumigant and shortens fumigation time. However, some materials may be damaged by vacuum and require special precautions. Atmospheric chambers are useful for fumigating materials that might be damaged in a vacuum chamber.

Tarpaulin Fumigation

Tarpaulin fumigation involves placing a gas-tight material over the commodity or structure to be fumigated. The tarps must be specially made for fumigation, such as impregnated nylon or sheet polyethylene. (Waterproof canvas tarpaulins are not satisfactory.) Polyethylene tarps can be used in thickness from four to six mils. Use gas-impervious adhesive tape to join various sections of polyethylene film.

The tarpaulin method provides thorough protection from insect damage at a practical cost. Done in place, it permits fumigation without the expense of moving huge stores of commodities. Tarpaulin fumigation can effectively and economically free materials such as bagged grain, dried fruit, stacked lumber, and other commodities from insects.

Tarpaulin fumigation may be done in the open, on loading docks, or in areas of buildings that allow safe aeration when the tarpaulin is removed. However, sites must also be checked for possible hazards in securing the fumigated area from humans and animals as well as for adequate sealing.

Rail-Car and Truck Fumigation

Items shipped in rail cars or in large truck trailers are often fumigated after they are loaded into the vehicle. This prevents pests from being transported to other locations and protects shipped products from pest damage during transport. Most vehicles, depending on their condition and on the type of commodity being fumigated, require tarping or other sealing to confine the fumigant.

Pests controlled by rail-car and truck fumigation include beetles and moths that infest flour, grains, nuts, dried fruits, and other agricultural products. These insects usually are brought into the vehicle on the commodity being shipped. Some insect pests may hide in empty vehicles, feeding on residues from previous cargoes. Unless controlled by fumigation or removed by thorough cleaning, these pests can infest future loads.

Fumigation of rail cars and truck trailers must comply with the regulations of state and local highway departments and departments of transportation as well as fumigant-label instructions. In some cases, loaded rail cars can be fumigated in transit. However, regulations prohibit truck trailers from being moved until fumigation and aeration have been completed. When performing a truck or boxcar fumigation, the pesticide applicator must post warning signs on all entrances to warn of the hazards.

Because fumigated boxcars or trailers may contain residues of a fumigant after aeration, the vehicles need to be monitored with appropriate detection equipment once they reach their destination and before they are unloaded. The person opening and monitoring fumigated loads must wear respiratory protection and any other protective equipment required by the fumigant label.

FACTORS AFFECTING FUMIGANT PERFORMANCE

For a fumigant to work effectively, the correct concentration of gas molecules must be present in the atmosphere surrounding the target pest. Molecule concentration may be affected by several factors. Some important ones are:

- 1. Sorptive quality of the treated commodity, either through absorbing (taking fumigant into the commodity) or adsorbing (fumigant condensing on the surface of the commodity).
- 2. Temperature and humidity during treatment.
- 3. Speed of diffusion of the fumigant through the commodity.
- 4. Reactions of the fumigant with other chemicals or articles in the treated area.
- 5. Amount of fumigant applied.
- 6. Susceptibility of the target pests.
- 7. In fumigation chambers, the pressure of the gas in the chamber.

Sorptive Qualities

Surfaces or items within the fumigated area may affect the concentration of fumigant molecules. For instances, cardboard boxes that contain produce or other food items will absorb some of the fumigant. Foam rubber used in upholstery or as carpet padding is also sorptive. Building insulation has large surface-areas and therefore will sorb fumigant molecules.

Molecules can be either absorbed or adsorbed. (See No. 1 in above list.) When absorbed, fumigant molecules dissolve into another material, such as water, oil, or other liquid. Absorption may not always be reversible, therefore resulting in greater problems with chemical residues.

Adsorption

Adsorption is a molecular attraction between gas molecules and the surface of something in the environment. The rate of adsorption is influenced by temperature. Fumigants applied while temperatures are low will adsorb more rapidly than when applied under higher temperatures. Adsorbed molecules may be released (desorption) as the temperature rises and as the concentration of the gas molecules in the surrounding atmosphere decreases. Fans and blowers that force air through the commodity can further speed the reversal of fumigant adsorption.

Temperature

Temperature at the treatment site affects both the fumigant and the target pest. Low temperature increases the sorption rate of the fumigant so that the concentration of the fumigant is reduced, but desorption is slowed by cooler temperatures. Fumigants also volatilize and diffuse more slowly at cooler temperatures.

Insects and other target pests may be less sensitive to effects of fumigants at lower temperatures. During cooler conditions, respiration of the target pests is slowed, making them less susceptible to poisons that affect respiration. Preferred fumigation temperatures usually range between 50 and 95 degrees F. Check the label of the fumigant being used for its optimum temperature and acceptable temperature range. Also remember that during the course of a fumigation application, the temperature of the treated area can

decrease or increase due to fluctuations in outside temperatures and also due to the cooling action of the fumigant being released (latent heat of vaporization).

Humidity

The concentration of water vapor in the atmosphere -humidity -- can affect the performance of fumigants that are water-soluble, such as methyl bromide. The watersoluble fumigants become unavailable when dissolved in water, reducing their concentration. Fumigants may not be able to penetrate wet areas, allowing insects in those areas to survive.

High humidity can also create moisture condensation in the fumigated area. Condensation can cause spotting of treated surfaces. In stored grains, condensation can cause wet spots that allow molds and storage heating to develop.

Diffusion

For a fumigant to be effective, it must penetrate the entire treatment site quickly and must be in the proper concentration. Factors that slow the diffusion rate include heavier fumigant molecules, low diffusion potential, and cool temperatures. Diffusion may also be hampered by dust in the fumigated area, a common problem in fumigating grain-storage structures. A fan or blower will increase diffusion.

Reaction with Other Chemicals

Materials in the treatment area, including food products being treated, may react chemically with a fumigant. Higher temperatures may further speed reaction processes. For instance, the flame from a pilot light or heat from a glowing electric heating element may cause fumigant molecules to react with other gas molecules in the air. Chemical reactions of this type are not reversible under normal conditions. If fumigant molecules react chemically, new chemical compounds will be formed. This may include corrosive acids, such as result from heating sulfuryl fluoride. Possible residues of newly formed chemicals may also stay in the fumigated area or on treated food products. For example, inorganic bromide compounds are found as residues on some food items that have been fumigated with methyl bromide.

In addition, chemical reaction of fumigants may lower the concentration of the fumigant enough to reduce the effectiveness of the fumigation. Check the fumigant label for precautions, and inspect the fumigation site thoroughly to eliminate materials or conditions that may allow reactions to occur.

Concentration and Time

How well a fumigant works depend both on the amount that has been applied and how well the concentration of gas molecules is maintained after application. The amount of fumigant applied is usually expressed in weight per volume (for example, pounds per 1,000 cubic feet or grams per cubic meter). The concentration of a fumigant is the amount of gas present in a given volume (for example, ounces per 1,000 cubic feet or milligrams per liter). Concentration is influenced by sorptive qualities, temperature, chemical reactions, and how well the fumigated area is sealed.

When fumigating grains, the applicator needs to adjust the dosage for the intergranular space (the amount of space between individual grains). This factor varies with the type and condition of the grain. Read the fumigant label for dosage information about these sites.

It's also important to maintain the critical amount of gas in the area of the target pests for a certain period of time. Although most fumigants are fast-acting, effective concentrations need to be maintained for several hours to days or weeks to allow control. For example, in order to kill 99 percent of the cadelle-beetle larvae in stored grain, a concentration of 33.2 milligrams per liter must be maintained for five hours.

Susceptibility of Target Organisms

Target organisms can react very differently to the effects of fumigants. This variation may be due to species differences. It can also be acquired, by development of populations that are genetically resistant to the treatment. Variation in susceptibility also is affected by the life stage of the pest.

In addition, the way a fumigant is applied can sometimes influence pests' susceptibility. For instance, some insects can tolerate a higher concentration of fumigant if they are first exposed to a low concentration for a short time. To avoid this problem, bring the fumigant level to the

lethal concentration quickly, then maintain the level throughout the fumigation period.

Pressure

In an airtight chamber, the penetration rate of fumigants may be controlled by using positive or negative (vacuum) pressure. Too much pressure or vacuum may cause structural changes in the commodity being fumigated.

Therefore, care must be taken to prevent damage to the commodities. To prevent undue expansion of tightly sealed packaged goods while a vacuum is being created, the pressure should be lowered slowly and/or the decompression process should be stopped for two to five minutes after each one-inch fall of mercury in a pressure-measuring device.

SAFETY PRECAUTIONS AND PROTECTIVE DEVICES

Using Two Trained Applicators

Recent regulations and changes in warning statements on labels now require the presence of two trained applicators during hazardous stages of fumigant application. This strengthens long-standing recommendations to always work in pairs. Two applicators are to work together whenever the application or gas-monitoring requires entry into or work within the confined space where a fumigant is applied.

Aluminum-phosphide and methyl-bromide labels do allow an applicator to work alone if the fumigant is applied outdoors to a moving grain-stream (aluminum phosphide) or in recirculation systems where methyl-bromide concentrations don't exceed five ppm in the work area. Even so, the presence of two trained applicators is always a wise investment for safety in the event of accident or emergency.

Exposure Levels

Respiratory protection is required for certain phases of most fumigant applications and other times when the airborne concentration exceeds a set value. The American Conference of Governmental Industrial Hygienists (ACGIH) recommends following established exposure limits known as **Threshold Limit Values** (TLVs). A TLV is the airborne concentration of a fumigant in parts per million that nearly all workers may be repeatedly exposed to on a daily basis without adverse effect. A TLV is usually established for each fumigant as a guide to prevent health hazards, but it should not be considered the distinction between safe and unsafe fumigant concentrations.

Two types of TLVs are recommended by the ACGIH as guidelines for protecting persons handling toxic vapors, including fumigants. These recommendations are not enforceable standards, although they contribute to better worker safety. The two recommended TLVs are:

- 1. TLV-TWA, the **threshold limit value-time-weight average**, is an airborne concentration, in parts per million, of a fumigant (or other toxic gas) that most workers can be exposed to during an eight-hour workday or 40-hour workweek without developing health problems. Typically, the TLV-TWA value is the concentration referred to on fumigant labels that must not be exceeded without appropriate respiratory protection.
- 2. TLV-STEL, the **threshold limit value-short-term exposure level**, is the maximum allowable concentration of any fumigant that a person should be exposed to without respiratory protection. It's recommended that exposure to this concentration be for no longer than **15 minutes** at a time, with a minimum of 60 minutes between exposure periods. No more than four exposure periods should be allowed in one workday. The total exposure for any single day should not exceed the TIV-TWA level for an eight-hour work period.

If the fumigant doesn't specify a maximum exposure value, exposures must then be kept below the Permissible Exposure Limit (PEL). PELs are set forth in state or federal health and safety regulations. These limits usually represent the maximum concentration of an airborne chemical that can be present without being a health hazard to most people.

The TLV and PEL values should only be used as a guide, since these levels may not protect everyone under all types of conditions. For example, there may be a few workers who will be sensitive to effects of the chemical below the TLV or PEL. Heavy physical activity, which increases the breathing rate, increases chemical uptake of airborne chemicals. Also, the exposure levels are based solely on exposure through inhalation. Since some fumigants can be absorbed through the skin or accidentally ingested, this increases overall exposure levels.

Gas-Detection Devices

Revised labels for fumigants require the use of sensitive gas-monitoring devices during fumigant application and before warning placards can be removed from fumigated storages.

Devices that provide adequate sensitivity includes detector tubes and matching pumps manufactured by Auer, Draeger, Matheson-Kitagawa, MSA and Sensidyne. Detector tubes are sealed glass tubes filled with a specific, reactive solid. Both ends of the tube are broken off just before use, and one end is attached to a calibrated pump. Available pumps use a bellows, bulb, or piston-type syringe to draw a precise volume of air through the detector tube. Discoloration of the solid material within the tube indicates fumigant concentration; gas concentrations can be read directly from the glass tube. Tubes and pumps manufactured by different companies may be very similar, but to get accurate readings, it's necessary to match detector tubes and pumps from the same manufacturer. Don't mix separate brands of equipment.

"Low-range" detector tubes that accurately indicate low levels of fumigant concentrations are required for label-specified monitoring practices that provide information for worker safety. "High-range" tubes may be useful for detecting fumigant leaks. These tubes are scaled for measuring much higher concentrations of fumigants, and they are especially useful for monitoring gas concentrations within storages during fumigation to determine if the necessary levels were reached. Other gas-monitoring devices, such as halide leak-detectors and thermal-conductivity meters, may be used to detect leaks or determine internal concentrations of gas during fumigation. However, these devices don't provide

label-required levels of sensitivity necessary for determining safety (respiratory-protection) needs. Halide detectors also should not be used around grain-storage buildings, since the flame may trigger an explosion of grain dust.

When measuring fumigant levels after fumigation, it's important to take readings from several locations. Often fumigants may become trapped in localized pockets. Different materials will also desorb at varying rates, a process called offgassing. This can allow toxic levels of the fumigant to occur in scattered locations.

Detector tubes are specific for a single fumigant. Auer, Draeger, Matheson-Kitagawa, MSA and Sensidyne manufacture detectors that offer adequate sensitivity for label-required monitoring of hydrogen phosphide (phosphine) and/or methyl bromide.

The only currently available detector that offers adequate sensitivity for label-required detection of chloropicrin is produced by Matheson-Kitagawa and is specified on the chloropicrin label.

Tubes available for measuring CO₂ concentrations are available from several manufacturers. Several types of tubes may be needed, since applicators must be able to measure low (below one percent) CO₂ levels to provide information on worker safety and high concentrations (up to 60 percent, minimum) to determine the need for continued injection of the gas into the structure.

Warning agents, such as chloropicrin, are sometimes added to fumigant gases that otherwise have little odor. Sometimes warning agents may affect the accurate reading of fumigant levels, so be sure to use detecting equipment that can reliably measure fumigant concentrations in the presence of the warning agent. (Note: Commodity fumigators must not use chloropicrin as a warning agent, since this material isn't permitted for use on most food items, and it would result in an illegal residue.)

Protective Equipment

Various types of respiratory equipment are available, but their effective and safe use requires that the equipment is matched to the specific need. For example, chemicalcartridge respirators, used in applying many types of pesticides, are not suitable for fumigation work. Depending on the gases and their concentration, various types of gas masks, self-contained breathing apparatus, or air-supplied systems are appropriate.

Canister and cartridge-type **gas masks** employ a replaceable canister or cartridge that contains chemical components that absorb specific gases. Full-face canister respirators (not half-face cartridge respirators) should be used as protection where this meets label specifications.

The effective life of an individual canister varies according to fumigant concentration and the respiratory rate of the applicator. Maximum limits are stated on each canister. Under NIOSH/MSHA regulations, canisters are color-coded according to fumigant absorbency. For example, canisters approved for protection from hydrogen phosphide are coded yellow with an orange stripe. Canisters effective for methyl bromide and chloropicrin are color-coded black. Always double-check the color code with written specifications that indicate the canister is effective for the fumigant.

As a canister <u>empties</u>, it becomes hot. Breathing hot air or encountering high resistance to breathing provides a warning that the canister is about to become ineffective for protection.

When this occurs, or when the applicator smells or tastes the fumigant or experiences poisoning symptoms, the applicator should **immediately** leave the fumigated area. An empty canister should be crushed before it's discarded so that no one will mistakenly use it in the future.

Canister respirators are inadequate for use in oxygen-deficient (less than 19.5 percent) environments, such as that produced by CO_2 fumigation. Although canisters may absorb the toxic fumigant in this environment, they don't supply necessary oxygen. In these situations, a self-contained breathing apparatus (SCBA) or a combination air-supplied/SCBA is needed.

General Rules on Canister Use

1. Discard any canister that has been used for more than 30 minutes (total time) in a fumigant atmosphere.

- 2. Discard any canister whenever an odor of fumigant is detected as coming through. (The absorption material isn't working).
- 3. Discard any canister used for less than 30 minutes if it's more than one year old.
- 4. Discard canisters with expired expiration dates or that have been manufactured more than two years earlier (even if unused), unless the instruction sheet specifically says otherwise.
- 5. DON'T use a canister-type gas mask to enter a recently fumigated or oxygen-deficient area.

The self-contained breathing apparatus (SCBA) common-ly used for fumigation is the air pack. Air packs comprise a full-face mask attached to a tank of air carried on the applicator's back. An air pack supplies up to 25 to 30 minutes of air supply and allows work in an oxygen-deficient environment. This time period may be considerably shorter if overexertion increases the rate of breathing. A warning bell can be set to signal the depletion of the air supply.

In a fumigant-laden storage, safe exits may require uninterrupted respiratory protection. For this reason, carrying an approved canister respirator when using an air pack is recommended for situations where oxygen concentrations remain adequate. The canister respirator will allow emergency escape if the SCBA expires or malfunctions.

Methyl-bromide and chloropicrin label directions concerning respiratory protection include reference to combination air-supplied/SCBA respirators. Air-supplied respirators employ an outside air source pumped to the applicator through an air line. The major advantage of the air-line system is that the air supply doesn't expire in a limited time. Disadvantages include the need to tow the air line throughout the storage. Air-pump failure or a constriction of the air line can shut off the air supply. The air pump must also be located in a fumigant-free area. In combination with an SCBA, an air-supplied respirator does offer an unlimited work period with backup respiratory protection provided by the SCBA, if for any reason the outside air supply is cut off.

One final respiratory-protection topic concerns the fit of a face mask. If a face mask doesn't seal tightly against the face, it cannot provide protection from a fumigant gas. An applicator must select a mask that fits his or her face; facial hair must be cleanly shaved to allow a tight fit. Fit can be tested by closing off the breathing tube and trying to breathe in or blow air out.

Air passage between the mask and the face indicates an unsatisfactory fit. The release of irritant gases near the edges of the mask can also indicate an improper fit. Detection of the irritant within the mask signals a poor seal between the mask and the face.

None of the respiratory-protection equipment provides protection from skin absorption or skin injury by fumigants. When using fumigants that have potential to injure the skin or be absorbed by it, such as chloropicrin, be sure to wear additional protective clothing.

Respiratory-Protection Requirements

Fumigant labels require the use of specified types of respiratory-protection equipment during most fumigant applications. Labels specify maximum fumigant concentrations in which applicators can work without respiratory-protection equipment. Gas concentrations greater than the label specifies signal a need for exposed workers to use respiratory-protection equipment. These levels include the following:

- ! Workers exposed to hydrogen phosphide (phosphine) at levels above 0.3 ppm must wear a canister-type gas mask or self-contained breathing apparatus (SCBA).
- ! Workers exposed to concentrations of hydrogen phosphide above 15 ppm, or where levels are not measured, must wear an SCBA.
- ! Workers exposed to chloropicrin concentrations above 0.1 ppm must wear a canister-type respirator, an SCBA, or a combination air-supplied/SCBA respirator.
- ! Workers exposed to methyl-bromide levels above five ppm must wear a SCBA or a combination air-supplied/SCBA respirator.
- ! Workers exposed to carbon-dioxide concentrations exceeding 1.0 percent must wear an SCBA or combination SCBA/air-supplied respirator.

Symptoms of Exposure to Fumigants

Most reactions to fumigant poisoning differ from those of exposure to other pesticides. For instance, many fumigant reactions simulate drunkenness. Symptoms of fumigant exposure can include:

- ! Slowed body movements
- ! Slurred/slowed speech
- ! Dizziness
- ! Numbness of hands or feet
- ! Coughing
- ! Sneezing
- ! Dryness/irritation of nose and throat
- ! Breathing difficulty
- ! Nausea
- ! Abdominal pain.

CALCULATING USE RATES

The first step in deciding how much fumigant must be released to achieve the desired concentration is to measure the length, width and height of the area to be treated and figure its volume. (Volume measurements on fumigant labels are given in cubic feet.) If the commodity, container or structure to be fumigated is to be tarped, the total volume inside the tarp must be determined, not just the volume of the structure or commodity. This must account for spaces caused by overhangs, eaves, and other irregular shapes.

Fumigant labels are the best source of information on calculating the proper amount of fumigant to use for specific situations. Always read and follow label instructions.

Physical or environmental conditions may influence the amount of fumigant that must be applied to achieve the required dosage. Factors that must be considered include:

- 1. Temperature and temperature fluctuation during fumigation.
- 2. Sorption qualities of the commodity or items in the target site.
- 3. The type and condition of the sealing method.
- 4. Texture and moisture content of the soil beneath the fumigation site.
- 5. Wind velocity during the fumigation period.
- 6. The volume of the area being fumigated. Some fumigant manufacturers furnish calculators, charts,

or slide rules to help figure dose adjustments for these factors.

Low temperatures may affect dispersal of the fumigant in a treated area. Also, insects are generally more resistant to a fumigant when the temperature is low because their metabolism slows. If a fumigation lasts over a lengthy period, there also may be important temperature fluctuations. Heating the fumigated area and increasing air circulation can overcome most low-temperature-related problems.

Sorption of fumigant molecules by commodities or surfaces in the treatment area reduces the concentration. Little can be done to change sorption qualities. Therefore, the applied dose of fumigant may need to be increased. Whenever fumigation takes place over soil, such as outdoor bulk-commodity fumigation or structure fumigation, the texture of the soil and its moisture content will influence fumigant concentration. Soils made up of fine clay or loam have less space between particles. Coarser soils, such as silt or sand, are much more porous and hold more fumigant. Increasing soil moisture by wetting will lower fumigant loss.

The way a fumigation site is sealed influences how fast fumigant molecules escape. Poorly sealed seams or holes in the tarp allow a quicker drop in concentration. Wind may increase the amount of air exchange between the fumigation site and the external environment, speeding fumigant loss.

As the size of the fumigated area increases, a smaller dose per unit of volume is usually needed to achieve the desired concentration.

SEALING

In most cases, the only way to achieve a sufficiently high concentration of fumigant is to seal the treatment area with a gasproof barrier. The best possible seal should be at the lower sections of the fumigated space, since most fumigants are heavier than air. Leaks in the lower portion of the fumigated space will allow more fumigant to be lost than leaks in upper areas.

Sealing can be accomplished in several ways. One method is to cover the treatment site with four- or

six-mil polyethylene sheeting or an impregnated gasproof tarpaulin. Some containers, such as storage bins, may be sufficiently airtight and may only require taping around openings or vents. Seams or cracks can sometimes be sealed with a liquid that expands to form a solid foam after being applied. Sealants that come into contact with food must be approved food-grade sealants. Fumigation chambers are built to be airtight, so they usually need no additional sealing.

It may be necessary to wet the soil around the foundation when doing structural or outdoor fumigations. This will reduce the amount of fumigant that will pass into the soil and will help achieve a good seal. Concrete or asphalt surfaces provide a satisfactory seal.

In structural fumigations, look for potential problems associated with tarping, such as landscape plants. Plants can interfere with sealing, and they may be damaged by the fumigant. Fragile roofs or roof-mounted structures can also be damaged in the sealing process unless special care is taken.

Structural fumigation may also require sealing of areas within the structure. Materials that may be damaged by the fumigation and that can't be removed should be sealed off to exclude the fumigant gas. Drains and other conduits for the fumigant may also need to be blocked. However, it should be recognized that these areas may also provide refuges and, if not exposed to the fumigant, allow some of the target pests to survive the treatment.

Seal seams and holes in the tarp with durable tape or clamps. Cover sharp edges of a vehicle, container, or structure with protective material, such as foam rubber, to keep them from tearing the tarp.

Moisture may condense inside the tarp or on surfaces or commodities being fumigated. Condensation is greatest during periods of high humidity or falling temperature. High humidity may be due to a recent rainfall or because of high humidity of the commodity. Condensation can cause several problems, such as interfering with the fumigant and damaging commodities by staining, spotting, and surface corrosion. If possible, dry out commodities or areas having high moisture before beginning a fumigation. Drying can be quickened by heating, exposing the materials to sunlight, increasing air

circulation, or -- in closed buildings -- running air conditioners before and during tarping to remove moisture.

With tarpaulin fumigation, careful consideration must be given to the method of obtaining a ground seal. If concrete and asphalt surfaces are smooth, they are satisfactory. Wood surfaces are not suitable. With wood and most soil surfaces, it's necessary to place a section of the tarp material beneath the commodity as well as over it.

There are several methods of getting a good ground seal. Allow enough tarp material to skirt outward to at least 18 inches from the stack. Loose sand, sand snakes, or water snakes can then be used to hold the skirt to the ground surface.

Occasionally, a stack may be too close to a wall to obtain a good ground seal. If the wall is well-sealed, the solution is to seal the tarp directly to the wall with adhesive tape.

APPLYING THE FUMIGANT

Before applying any fumigant, notify local fire and police authorities and other security personnel as to the location, the chemicals to be used, proposed date and time of the fumigation, type of protective equipment required, and fire hazard rating. If necessary, provide authorities with pertinent safety literature on the materials to be employed. In addition to normal equipment needs, also arrange for standby equipment, replacement parts, and an alternate plan of action.

The release of a fumigant into an enclosed area is referred to as "shooting" or "shooting the fumigant." Methods of application, or shooting, vary according to the type of fumigant used, what is being fumigated, and where the fumigation takes place. The way fumigants are applied in any situation, however, influences the degree of control of the target pests. Incorrect application techniques can damage the area, damage the commodity, or injure people.

Gas Fumigants

Gas fumigants come packaged under pressure in large steel gas-cylinders or small metal cans. When using fumigant from a large cylinder, suspend the cylinder from a scale and monitor its weight change over time to calculate the rate of application. The total weight of fumigant used will determine the dosage applied.

Gas is injected into the treatment area through one or more hoses or shooting tubes. Rate of application is influenced by the diameter and length of the shooting hose. Nozzles attached to the shooting hose further affect the fumigant release rate. Cylinder pressure also controls the rate of release, the pressure in the cylinder being influenced by the remaining gas and cylinder temperature. Obtain charts from the fumigant supplier to calculate the optimum release rate for the fumigant being used.

Releasing fumigant too fast may cause rapid cooling of the fumigation site and result in poor fumigant distribution. Rapid cooling will also promote condensation of water vapor. Releasing the fumigant too slowly may cause icing of the shooting tube and possibly restrict the flow of fumigant. As the ice melts, it may spot or stain. Slow release may also prevent the fumigant from reaching the effective concentration quickly enough to control the target pests.

Fans or blowers should always be used when the fumigant is heavier than air. Continue the use of fans or blowers until the desired concentration of fumigant is achieved uniformly throughout the fumigated space. Discontinue their use after this point to reduce potential leaking.

Pelleted Fumigants

Aluminum phosphide for use in protecting bulk grain needs to be evenly distributed to provide adequate fumigant levels. Pellets should be inserted deeply within the grain mass, at least five feet, and no more than 50 pellets or 20 tablets should be inserted per probe. Applicators should also wear cotton gloves so that perspiration doesn't contact the aluminum phosphide, releasing the phosphine gas. The applicator not making the probes should periodically monitor fumigant (hydrogen-phosphide/phosphine) levels.

Since hydrogen-phosphide gas doesn't provide adequate and uniform concentrations more than 30 feet below its application site, supplemental insertions may be needed in larger structures. These may sometimes be able to be inserted through the aeration or drying fan. Within grain-storage facilities, it's best to tarp over the surface of the grain mass. If the grain isn't tarped, the fumigant rate must be increased to provide adequate concentration in the bin headspace as well as within the grain mass. Eaves and roof hatches must be tightly sealed if the grain surface isn't covered.

Liquid Fumigants

Liquid fumigants volatilize rapidly into a gas. The gas then penetrates through the commodity being treated. To assure even distribution, apply liquid sprays to commodities as they are being loaded into a storage container or storage building. Consult the fumigant label to determine protective clothing and equipment needs for these applications.

Carbon Dioxide

Use of carbon dioxide as a fumigant requires special application, since the gas must displace much of the existing air to achieve the necessary concentration, usually 60 percent. This requires introducing large amounts of the gas and venting the structure to allow the normal atmosphere to be expelled. A top-down purge involves allowing the CO_2 to be introduced at the top of the structure, displacing air as it settles downward. Bottom injections of carbon dioxide are sometimes used in storages with leaky roof or eave areas. Following the purge, complete sealing will result in the most successful fumigation. Additional injections of small amounts of CO_2 should be made when measurements indicate that concentrations have dropped below 50 to 60 percent.

POSTING AND SECURING A FUMIGATION SITE

Before fumigating, fumigated areas must always be clearly posted to direct others to stay away. Individual fumigants include detailed instructions for the posting of warning placards on fumigated structures. Labels specify the wording (including some information in both English and Spanish) and content that must appear on warning placards. Placards must be placed around the perimeter of the treatment area and at all entrances.

On structures, all entrances should be locked during fumigation and access allowed only to authorized persons, and even then only in an emergency. Use secondary locks on all doors to further guard against unauthorized entry.

Areas that cannot be locked or secured must have someone present throughout the fumigation and aeration period to block unauthorized entry. Always make sure to inform janitors, watchmen, and other persons who regularly use the building about the fumigation.

Warning placards may not be removed and the commodity may not be processed or fed until a certified applicator uses an appropriate gas-detection device to determine that gas concentrations have dropped below specified levels for the fumigant. This follow-up monitoring is a practice recently required under newer regulations.

AERATING

When fumigation is completed, the fumigant must be completely dissipated by aeration before allowing access to anyone or before vehicles can be moved.

Bulk Grain and Other Commodities

Wear respiratory equipment to aerate bulk items that have been covered with a tarpaulin. First, pull the tarp up from the sides for about 30 minutes, then remove the tarp completely. If this fumigation was made inside a building, open doors and windows and use fans to exhaust the fumigant. The air being exhausted from the building must be directed away from work areas, sensitive plants, and neighboring property. Make sure downwind areas in the vicinity are kept clear to prevent people or animals from contacting the fumigant as it disperses.

Wear respiratory protection, if entering the building. Check the fumigant level to determine the level at which it's safe to allow re-entry without protective equipment. Take measurements at several locations.

Continue to keep people away until monitoring equipment confirms that the fumigant level is below the harmful stage. Continue to aerate for several hours, and leave the building or commodity unsealed; this will prevent a fumigant buildup by desorption.

Vehicles

Roll back the tarp that covers boxcars or truck trailers, and open the doors and ventilators for 30 to 60 minutes; then remove the tarps. Measure the fumigant level before entering the fumigated area, and wear protective equipment until the fumigant has dispersed. To prevent injury, notify people who will open the vehicle at its destination that they must wear respiratory protection.

Instruct them not to unload the vehicle until fumigant levels have been monitored again and are determined to be in the safe range.

Buildings

Wear the label-recommended respiratory protection when beginning the aeration of structures. Use fans to force the fumigant out of the structure. Fans should be installed during tarping and before the fumigant is applied. Connect electrical cords to a remote power source so the fans can be started without entering the fumigated area. Be sure the exhaust from the fans is directed away from work areas, sensitive plants, and neighboring work areas.

Begin by starting the fans, which will pull the tarps up against the sides of the structure. When the tarps have drawn up tight, slightly open a seam on the opposite side of the building from the fans. Wearing respirator equipment, enter the structure and close outside doors and most windows to prevent fumigant inside the building from getting into the space between the tarp and the building's exterior.

Next, remove the bottom seal, working in both directions away from the exhaust fans. Open all lower tarp seams before opening roof seams. Pull tarps up or peel them away from the sides of the building rather than dropping them to the ground. As soon as tarps are being removed, the exhaust fans can be shut off and all doors and windows opened.

After tarps have been removed, use an atmospheremonitoring device to determine when the fumigant has been dissipated well. Check for pockets of the fumigant in low areas and in corners, closets, or other areas where there is poor air circulation. Areas where there are porous materials may have sorbed quantities of fumigant that will be more slowly dissipated. Longer aeration periods will be needed under these conditions. To hasten the desorption process, increase the temperature inside the structure and maintain good air circulation. Be sure to ventilate refrigerators and freezers as well as attics and crawl spaces.

SPECIAL CONSIDERATIONS FOR FUMIGANT AND CONTAINER DISPOSAL

Fumigants are hazardous materials; empty containers and any excess or unused fumigants must be handled appropriately. The correct method of handling fumigant containers differs among fumigant products. Empty canisters or tanks that were used to hold methyl bromide or CO₂ under pressure should be returned through the original shipper to the manufacturer for refilling. Some chloropicrin containers should be handled in the same manner, while others should be triple-rinsed, punctured, and disposed of in landfill. Consult specific container labels for instructions.

Unused aluminum-phosphide pellets or tablets in opened flasks should not be disposed of. Once flasks are resealed, these tablets or pellets can be stored safely (as long as the label remains intact) for future use. Don't store flasks at sub-zero temperatures, as doing so will increase the likelihood of ignition (flash) when they are opened.

If aluminum-phosphide tablets or pellets are spilled or flasks are punctured, hydrogen-phosphide gas is released. Persons cleaning up the spill or working in the contaminated area must wear an SCBA unless gasdetection equipment is used. If gas concentrations are measured and if hydrogen-phosphide concentrations range between 0.3 and 15.0 ppm, a canister respirator should be used. At higher concentrations, an SCBA is required. Cleanup personnel should wear cotton or neoprene gloves while handling spilled material. If a spill can be cleaned up immediately, spilled pellets or tablets should be used immediately or transferred to an empty

flask with an intact label. If such a container isn't available, tablets or pellets can be placed in a sound, DRY metal container that should be sealed and labeled as aluminum phosphide. Keep the original product label with the substitute container.

If spilled material has begun to react and decompose, or if it's contaminated by other substances so that it can't be safely stored, it should be gathered and placed into open-top, perforated gallon cans and processed immediately. Don't use water to clean up an aluminum-phosphide spill.

Water will react with tablets or pellets to rapidly release hydrogen-phosphide gas, and the rapid production of gas can result in spontaneous ignition and explosion.

To deactivate unreacted or partially reacted pellets, transport them by hand or in an open vehicle to a location in the open air away from occupied structures. Fill a drum two-thirds full of water, and add one-fourth cup of low-sudsing detergent or surfactant for each gallon of water. Mix each flask of tablets or pellets with no less than one gallon of the water-detergent mixture. Wear respiratory-protection equipment, and slowly add the aluminum-phosphide product to the drum while stirring. Stir occasionally thereafter for at least 36 hours. DON'T COVER THE CONTAINER! Covering the container will confine the hydrogen-phosphide gas that is generated, and the resulting high concentrations may explode. This wet method of deactivation is preferred when five or more flasks of materials must be deactivated. The resulting slurry may be disposed of at an approved landfill.

An alternative to slurry deactivation is dry deactivation (for small quantities not exceeding five flasks). Pellets or tablets can be spread out in an open, secure area away from occupied buildings and deactivated by atmospheric moisture.

Disposal of residual dust from reacted pellets or tablets is necessary following a space fumigation. Residual dust is grayish-white, and it contains a small amount of unreacted aluminum phosphide. (Tablets or pellets that are only partially reacted remain slightly greenish in color and should be disposed of in the manners described above for spills.) Residual dust from up to five

flasks can be disposed of by on-site burial or by spreading over the land surface in a secure area away from inhabited buildings. This amount of dust may also be disposed of at a sanitary landfill or an approved pesticide incinerator. For larger amounts of residual dust, a detergent slurry disposal method, described above, is recommended. See product labels for additional directions.

Residual dust from up to three flasks can be held in an open one-gallon bucket pending disposal. Larger amounts of dust should be held in a porous cloth bag during any storage or transport before disposal. Don't put the residual dust from more than eight flasks of tablets or ten flasks of pellets in any one bag before disposal. Greater amounts may generate enough gas to risk explosion. Don't pile bags. Don't confine, dispose of, or store residual dust in closed containers such as dumpsters, drums, or plastic bags. Don't dispose of dust in toilets.

Empty flasks that contained aluminum phosphide may be recycled or disposed of in a landfill after they have been properly processed. To adequately clean flasks before disposal, flasks and stoppers may be triple-rinsed and then punctured. A small number of empty flasks may be punctured and held outdoors in an open and secure area away from occupied buildings to allow complete reaction of aluminum phosphide. Where triple-rinsing is used, rinsate may be disposed of in a landfill.

USE COMMON SENSE

It's essential that fumigators understand and follow the technical instructions that promote safe and effective fumigation of stored grain. It's just as important that fumigators remember to use good common sense when planning and carrying out a fumigation. Although it may be impossible to "teach" good common sense by writing instructions in study materials, the following comments are offered as reminders to exercise good judgement and to think ahead.

C Read and understand label directions. Demand information from the manufacturer and distributor. Don't use a fumigant without adequate training and confidence in your ability to do the job properly.

- C Supply local medical personnel with fumigant and poison-treatment information before using the fumigant.
- C Preplan the entire job. Think through every step, and plan your reactions to possible problems and emergencies.
- C Always work in pairs.
- Use, or have available, proper safety equipment. Make sure all equipment fits well and that all applicators are trained in and familiar with the use of necessary safety equipment.
- C Don't take shortcuts; follow through with well-planned and thorough application practices.
- C Don't become complacent. Each job is a new challenge and a new situation in which an emergency may require rapid and proper reaction.

THREATENED AND ENDANGERED SPECIES

The Endangered Species Act (ESA) was passed by Congress to protect certain plants and wildlife that are in danger of becoming extinct. This act requires EPA to ensure that these species are protected from pesticides.

Formulation of the Utah Threatened and Endangered Species/Pesticides Plan is a cooperative effort between federal, state, and private agencies and producers/user groups, and is a basis for continuing future efforts to protect threatened and endangered species from pesticides whenever possible. Furthermore, this plan provides agencies direction for management policies, regulations, enforcement and implementation of threatened and endangered species/pesticide strategies.

EPA has therefore launched a major new initiative known as the Endangered Species Labeling Project. The aim is to remove or reduce the threat to threatened and endangered species from pesticide poisoning. EPA has the responsibility to protect wildlife and the environment against hazards posed by pesticides. The ESA is administered by the U.S. Fish and Wildlife Service (FWS) in the U.S. Department of Interior. The Fish and Wildlife Service will determine jeopardy to threatened and endangered species and report to EPA. EPA and FWS will work cooperatively to ensure that there is consistency in their responses to pesticide users and to provide necessary information. The Utah Department of Agriculture is acting under the direction and authority of EPA to carry out the ESA as it relates to the use of pesticides in Utah.

Maps will show the boundaries of all threatened and endangered species habitats in affected counties. The maps identify exactly where, in listed counties, use of active ingredients in certain pesticides is limited or prohibited. Product labels will be updated as necessary. The updated labels will reflect any additions or deletions to the project. Because EPA's approach to the protection of threatened and endangered species was in the proposal phase at the time this guide was published, any and all of the above information on threatened and endangered species is subject to change and may not be valid.

WORKER PROTECTION STANDARDS

This final rule, which was proposed in 1988 and that substantially revised standards first established in 1974, affects 3.9 million people whose jobs involve exposure to agricultural pesticides used on plants; people employed on the nation's farms; and in forests, nurseries and greenhouses. The standard reduces pesticide risks to agricultural workers and pesticide handlers. The standard is enforceable on all pesticides with the Worker Protection Standard labeling. The provisions became fully enforceable in January 1995.

Agricultural workers in Utah now have a far greater opportunity to protect themselves, their families and others. These workers will know, often for the first time, when they are working in the presence of toxic pesticides, understand the nature of the risks these chemicals present, and get basic safety instructions.

Among the provisions of the rule are requirements that employers provide handlers and workers with ample water, soap and towels for washing and decontamination and that emergency transportation be made available in the event of a pesticide poisoning or injury. The rule also establishes restricted-entry intervals -- specific time periods when worker entry is restricted following pesticide application -- and requires personal protection equipment (PPE) for all pesticides used on farms or in forests, greenhouses and nurseries. Some pesticide products already carry restricted re-entry intervals and personal protection equipment requirements; this rule raised the level of protection and requirements for all products.

Other major provisions require that employers inform workers and handlers about pesticide hazards through safety training, which handlers have easy access to pesticide-label safety information, and that a listing of pesticide treatments is centrally located at the agricultural facility. Finally, handlers are prohibited from applying a pesticide in a way that could expose workers or other people.

GROUNDWATER CONTAMINATION BY PESTICIDES

Utah has implemented a comprehensive and coordinated approach to protect groundwater from pesticide contamination.

Formulation of the Groundwater/Pesticide State Management Plan is a cooperative effort between federal, state, and private agencies and producers/user groups; it provides a basis for continuing future efforts to protect groundwater from contamination whenever possible. Furthermore, this plan provides agencies with direction for management policies, regulations, enforcement and implementation of groundwater strategies.

While it is recognized that the responsible and wise use of pesticides can have a positive economic impact, yield a higher quality of crops, enhance outdoor activities, and give relief from annoying pests, the Utah Department of Agriculture is authorized by the U.S. Environmental Protection Agency (EPA) to enforce the protection of groundwater from pesticides. Product labels will be updated as necessary.

The Utah Department of Agriculture, in concert with cooperating agencies and entities, admonishes strict compliance with all pesticide labels, handling procedures and usage to protect groundwater in the state.

Groundwater can be affected by what we do to our land. Prevention of groundwater contamination is important, because once the water is polluted, it's very hard and costly to clean up. In some instances, it's impossible, especially if it is deep underground. City and urban areas especially contribute to pollution because water runoff that contains pesticides runs into drainage

tunnels, then into a river or an underground stream that drains into the river. For more complete information about what groundwater is and where it comes from, read the study manual "Applying Pesticides Correctly." Shallow aquifers or water tables are more susceptible to contamination than deeper aquifers. Sandy soils allow more pollution than clay or organic soils, because clays and organic matter absorb many of the contaminants.

The Federal Insecticide, Fungicide and Rodenticide Act (FIFRA), as amended, establishes a policy for determining the acceptability of a pesticide use or the continuation of that use, according to a risk/benefit assessment. As long as benefits outweigh adverse effects, a pesticide can be registered by the EPA. Although the intent of a pesticide application is to apply the pesticide to the target or pest, part of the pesticide will fall on the area around the target or pest. Rain or irrigation water then can pick up the part that is not degraded or broken down and carry it to the groundwater via leaching.

The major factors that influence the amount of contamination that can get into water are the chemicals' persistence in soil, retention time or time it remains in the soil, the soil type, the time and frequency of the application(s), soil moisture, placement of the pesticide, and the ability of the chemical to persist once in the aquatic environment. Each of these factors will influence the amount of pesticide that can leave the root zone or soil surface and percolate to groundwater.

Although some pesticides may have a high absorption quality, when they are applied to sandy soil, they will still migrate to the water table because there are no fine clay particles or organic matter to hold them. The management and use of pesticides is up to the individual applicator and/or land owner as to whether safe practices are used. Water is one of our most valuable resources; we must keep it as pure as possible.